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Aug 27, 2011 Stoichiometry - Study Guide" Given the equation $3\text{Fe} + 4\text{H}_2\text{O} \rightarrow \text{Fe}_3\text{O}_4 + 4\text{H}_2$ Use your answer from the example above as the theoretical yield.

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ANSWER KEY. synthesis $2\text{Sb} + 3\text{I}_2 \rightarrow 2\text{SbI}_3$. single replacement $2\text{Li} + 2\text{H}_2\text{O} \rightarrow 2\text{LiOH} + \text{H}_2$. decomposition $2\text{AlCl}_3 \rightarrow 2\text{Al} + 3\text{Cl}_2$. Stoichiometry Study Guide & Review

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